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[Electron - Wikipedia](#)

Hydrogen atom orbitals at different energy levels. The more opaque areas are where one is most likely to find an electron at any given time.

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The potential difference between electron acceleration in the ionisation source and electron ionisation. The electron energy for standard electron ionisation mass spectra is 70 eV, a level which maximises ion production and provides reproducible mass spectra.

[Electronvolt - Wikipedia](#)

In physics, the electronvolt (symbol eV, also written electron-volt and electron volt) is a unit of energy equal to approximately 1.6 10⁻¹⁹ joules (symbol J) in SI units.

[Energy Levels in Atoms - SDSS SkyServer DR14](#)

Electrons in a hydrogen atom must be in one of the allowed energy levels. If an electron is in the first energy level, it must have exactly -13.6 eV of energy.

[electron energy level - an overview | ScienceDirect Topics](#)

Iain R. McNab, in Encyclopedia of Spectroscopy and Spectrometry, 1999. Asymmetric top molecules. The energy levels of an asymmetric top molecule cannot in general be expressed in a simple algebraic form.

[Elektronenger te, prinzipien und systematik, \(Book, 1941 ...](#)

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